

Getting Energy Straight

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Force -- Newton -- Mass times Acceleration ( F = MA )
    Kilograms times Meters per Sec^2: (Kg x M) / S^2
        1 newton = 10^5 dynes
        1 pound-force ~= 4.5 newtons

Work -- Joule -- Force times Distance ( W = FD )
    (aka Energy)      Newtons times Meters -- N x M:      (Kg x M^2) / S^2
        1 Joule = 10^7 ergs
            .74 foot-pounds
            6.25x10^18 electron volts
        1 BTU = 1 Kilo-joule

Power -- Watt -- Work per Time ( P = W/S )
    Joules per Second -- J/S:      (Kg x M^2) / S^3
        1 HP = 550 ft-lb/s = 745.7 watts
        1 Kw = 1.34 HP
        1 BTU/hour = .29 watts

note:
    Watt = volt x ampere
        Volt = joule / columb
    1 Columb -- amp-sec ~= 6.25 x 10^18 electron-second
    Watt-seconds -- volt x columb
    1 Joule = 1 Watt-second
    1 KwHr = 3.6 Mega-joule

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for extra credit:

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Pressure -- Pascal -- Force per Area ( P = F/A )
    Newtons per Meter^2 -- N/M^2:      Kg / (M x S^2)
        1 pound/sqin (PSI) = 6.9 Kpascal
        1 dyne/cm^2 = 10 Pascal(/M^2)

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Or more succinctly:

distance(d)			
time(t)			
mass(m)			
velocity(v)		= d/t	
acceleration(A)		= v/t	= d/t^2
force (F)	= Am	= vm/t	= dm/t^2
work(W)	= Fd	= vmd/t	= d^2m/t^2
power(P)	= W/t	= Amd/t	= vmd/t^2 = d^2m/t^3

In the SI system, there are seven fundamental units:

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kilogram -- mass
meter -- length
candela -- luminous intensity (weighted to eye), 1/683-watt / steradian
second -- time
ampere -- current flow, columb per second
kelvin -- temperature (energy?)
mole -- Avogadro's # elementary entities == 6.02214179 x 10^23

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